pubs.acs.org/joc

Access to Substituted Dihydrothiopyrano[2,3‑b]indoles via Sequential Rearrangements During S‑Alkylation and Au-Catalyzed Hydroarylation on Indoline-2-thiones

Mukund Jha, $*$,[†] Ganesh M. Shelke,^{†,‡} T. Stanley Cameron,[§] and Anil Kumar[‡]

† Department of [Bi](#page-5-0)ology and Chemistry, Nipissing University, North Bay, ON P1B 8L7, Canada ‡ Department of Chemistry, Birla Institute of Technology and Science, Pilani, Rajasthan 333031, India § Department of Chemistry, Dalhousie University, Halifax, NS B3H 4J3, Canada

S Supporting Information

ABSTRACT: An efficient methodology for the synthesis of indole-fused dihydrothiopyrans has been developed from indoline-2 thiones. The protocol involves the synthesis of conjugated ene-yne-substituted indole-sulfides, a gold(III)-catalyzed rearrangement of the ene-yne side chain followed by intramolecular hydroarylation via C3−H functionalization of the indole core. This new synthesis of functionalized tricyclic indole derivatives through sequential rearrangements is quite general in nature

Transition-metal-catalyzed hydroarylation reactions have received considerable attention in recent years as an atomeconomical approach for the functionalization of arenes, as well as for the creation of complex molecular architectures.¹ This ever-developing area has witnessed a remarkable growth in gold-catalyzed methodologies involving the addition of [a](#page-5-0) wide variety of nucleophiles (C/N/O) to internal or terminal alkynes in an intra/intermolecular fashion.² Not surprisingly, this strategy has also emerged as a highly useful tool for the synthesis of novel heterocycles.³

Being structural constituents of several natural products, agrochemicals, pharmaceuticals[,](#page-6-0) and organic materials, heterocycles with diverse substitution patterns are highly sought-after chemical entities. In particular, chemicals possessing an indole core are an extremely desirable subclass of heterocycles due to a wide range of biological activities associated with them.⁴ A survey of the literature suggests the chemistry of S-containing heterocycles, thiopyran and fused-thiopyrans, has not [be](#page-6-0)en explored to the same extent as that of their pyran analogues. Recent reports have associated substituted thiopyran and fusedthiopyran scaffolds to anti-inflamatory, antibacterial, antihyperplasia, antipsychotic, analgesic, estrogen receptor modulator, and anticancer activities.⁵ Certain thiopyran-fused indoles and their pharmaceutically acceptable salts have been shown to possess psycho[an](#page-6-0)aleptic and nootropic effects.⁶ Considering the importance of functionalized indoles, developing an efficient synthesis process to access them remains a[n](#page-6-0) intense area of investigation. In this communication, we report a C−H activation-based two-step synthesis of dihydrothiopyrano[2,3 b]indole ring system starting from indoline-2-thiones. The newly generated indole-fused thiopyran derivatives are structural analogues of a biologically active carbazole skeleton.⁷

In a continuation of our efforts toward developing synthetic strategies to access [d](#page-6-0)iversely substituted indoles,^{8,9} we explored the application of 2-(2-alkenylthio)indoles to the construction of indole-fused S-containing ring structures. W[e h](#page-6-0)ave recently reported an efficient process for chemoselective S-benzylation of indoline-2-thiones using benzyl alcohols in the presence of a catalytic amount of Lewis acids, leading to biologically relevant indole-based sulfides.⁹ Mechanistically, the methodology supports the involvement of a resonance-stabilized benzylic carbocatio[n](#page-6-0) in the S-benzylation process.⁹ On the basis of these results, we envisaged that unsaturated aliphatic alcohols capable of producing resonance-stabilized ca[r](#page-6-0)bocations may also undergo S-alkylation similar to the S-benzylation process to produce 2-(2-alken/ynylthio)indoles. Furthermore, an intramolecular hydroarylation carbocyclization with the 3-position of the indole core may lead to an indole-fused S-containing ring system. Herein, we wish to report an efficient two-step methodology for the synthesis of indole-fused dihydrothiopyrans starting from indoline-2-thiones via sequential rearrangements (Scheme 1).

To the best of our knowledge, the use of unsaturated aliphatic alcohol[s](#page-1-0) in the synthesis of S-alkenylated sulfides has not yet been reported. Hence, our initial investigation focused on the S-alkylation of indoline-2-thiones with the aid of unsaturated aliphatic alcohols. Inspired by our S-benzylation work,⁹ we first employed rare earth metal triflates as a potential

Recei[ve](#page-6-0)d: November 13, 2014 Published: April 20, 2015

Scheme 1. Synthesis of Indole-Based Sulfides Using Substituted Benzyl Alcohols (Previous Work, Ref 9) and Synthesis of Indole-Fused Dihydrothiopyrans Starting from Indoline-2-Thiones via Sequential Rearrangements (Current Work)

Scheme 2. BF₃ Etherate-Mediated Alkylation of Indoline-2-thione (1a) using 2-Methylbut-3-en-2-ol (2a) and Prenyl Alcohol (2d)

catalyst for the S-alkylation of indoline-2-thione (1a) with 2 methylbut-3-en-2-ol (2a). Even after multiple attempts, the desired S-(2-alkenyl)indole could not be obtained under these conditions. Gratifyingly, the use of excess $BF₃$ etherate in the reaction facilitated the formation of desired indole sulfide 3aa in good yield. Interestingly, a rearrangement in the allyl side chain was observed in the isolated product (Scheme 2). It appears that the thermodynamically more stable, highly substituted alkene is the favored product under these conditions.

Encouraged by these results, we then proceeded to screen a range of substituted/unsubstituted allylic or propargylic alcohols with varying degrees of substitution $(1^{\circ}/2^{\circ}/3^{\circ})$ alcohols) for the S-alkylation reaction. As presented in Table 1, only a select group of unsaturated alcohols, primarily substituted allyl alcohols, yielded S-(2-alkenylated) products [\(](#page-2-0)3). No reaction was observed in the cases of unsubstituted allyl/propargyl alcohols. Interestingly, the regioisomer of 2a, prenyl alcohol (2d), when reacted with thione 1a, also resulted in 2-(3-methylbut-2-enylthio)-indole (3aa) in 69% yield (Scheme 2). These results were further corroborated using isomeric alcohols 2e and 2f, both of which resulted in sulfide 3ae in ∼50% yield. Furthermore, irrespective of the degree of substitution of the reactant alcohol, we observed a consistent trend of rearrangement in the reactions of all of the substituted allyl alcohols, including 3-methylpent-1-en-4-yn-3-ol (2g) and 1-ethynylcyclohex-2-enol $(2h)$ ¹⁰ under these conditions, in the synthesis of sulfides 3aa−ih (Table 1). The sulfides obtained from alcohols 2g and 2h are [of](#page-6-0) particular importance because they possess a conjugated ene-yne [un](#page-2-0)saturated system, which can be synthetically manipulated for further chemical transformations.

After acquiring all of the desired 2-(2-alkenylthio)indoles (3aa−ih), we attempted the metal-catalyzed hydroarylation

reaction to produce the indole-fused S-containing cyclized products. Initially, 2-(3-methylpent-2-en-4-ynylthio)-N-methylindole 3eg was chosen as a model reactant to study the intramolecular cyclization reaction with the prospect of producing annulated products I and/or II as depicted in Scheme 3.

As shown in Table 2, the desired C−H activated carbocy[cli](#page-2-0)zation step was screened using a range of potential metal catalysts in toluene [a](#page-2-0)t 80 °C. Initially, only reactions attempted with AuCl, AuCl₃, and $ZnCl₂$ resulted in product formation (Table 2, entries 1, 5, and 7). ^{1}H and $^{13}\tilde{C}$ NMR analysis of the product (liquid) revealed that the possible structure of the p[ro](#page-2-0)duct may be dihydrothiopyrano $[2,3-b]$ -Nmethylindole. The exact positions of the substituents were established using 1D NOESY experiments. The two possible regioisomers of 4a are shown in Figure 1 (A and B). The perturbations of the phenyl proton at position 5 (red dot) showed a strong NOE to the methyl a[nd](#page-2-0) vinyl groups at position 4 (blue dot). This evidence supported regioisomer A. In addition, negative evidence was obtained by the perturbation of N -CH₃ protons (black dot), which showed only an NOE response to the phenyl proton at position 8 of the skeleton. Thus, on the basis of 1D NOESY experiments, the structure of 4a was unambiguously assigned as 4,9-dimethyl-4-vinyl-4,9 dihydrothiopyrano[2,3-b]indole (regioisomer A).

Once the isolated product was fully characterized, the effect of a cocatalyst on the product yield was studied. On the basis of literature reports,¹¹ a combination of AuCl(PPh₃) and AgOTf was employed as a catalyst for our desired cyclization. However, this isolated yiel[d o](#page-6-0)f 4a was found to be lower than that of AuCl₃ alone (Table 2, entry 16). These results led us to select AuCl₃ as the catalyst of choice for the cyclization step and set out to optimize othe[r](#page-2-0) reaction parameters, such as temperature, solvent, and catalyst concentration. A loss in yield of 4a was

5273

^aReaction conditions: 1 (1.0 mmol), 2 (1.0 mmol), BF_3 etherate (2.0 mmol), dichloromethane (40 mL), rt; yields are based on the amounts of starting materials utilized in the reaction and are unoptimized.

observed when the reaction temperature was increased to 110 °C (Table 2, entry 8). Changing the solvent of the reaction did not appear to have a significant impact on the product yields. In general, reactions performed in toluene provided superior results among the tested solvents (Table 2, entries 9−12). Catalyst loading had a profound impact on the reaction outcome. The use of 2 mol % of $AuCl₃$ resulted in 50% yield of cyclized product 4a, whereas the use of 10 mol % of $AuCl₃$ provided 76% of 4a (Table 2, entries 5, 13, and 14). Overall, after a thorough screening, we determined that 10 mol % of

Table 2. Optimization of Metal-Catalyzed Hydroarylation Reaction^a

17 $AuCl_3(10)$ toluene 80 76 ^aReaction conditions: **3eg** (40 mg, 0.16 mmol), catalyst (2–10 mol %), solvent (2 mL), 80 °C, 12 h. ^bNo reaction.

Figure 1. Possible regioisomers of compound 4a.

AuCl₃ in toluene at 80 $^{\circ}$ C is the optimal condition for the synthesis of 4a from 3eg.

Next, the optimized conditions were used to study the scope of the C−H activation-mediated carbocyclization reaction. As shown in Table 3, differently substituted 2-(2-alkenylthio) indoles (3) were subjected to these conditions to obtain diverse indole-fused dih[yd](#page-3-0)rothiopyrans (4). Among the various substrates studied, no specific substitution effects were observed in the reaction outcome, and all dihydrothiopyran derivatives were prepared in good yields (54−76%). Owing to the nucleophilic nature of position 1 of the indole core in the case of unsubstituted 2-(2-alkenylthio)indoles (e.g., 3bg), a competition between position 1 and 3 was anticipated in the

Scheme 3. Chemical Structures of the Expected Annulation Products from Sulfide 3eg

Table 3. Synthesis of Dihydrothiopyrano $[2,3-b]$ indoles from 2- $(2-A$ lkenylthio)indoles^a

^aReaction conditions: 3 (0.16 mmol), $AuCl_3$ (10 mol %), toluene (2) mL), 80 °C.

cyclization process. In the past, gold-catalyzed reactions of indoles have been preferred using N-alkylated substrates to circumvent such issues.¹² Remarkably, no interference of indole nitrogen was observed in the current investigation. The synthesis of N-unsu[bs](#page-6-0)tituted analogues of indole-fused dihydrothiopyrans (4e−f) was achieved directly without requiring additional protection and deprotection steps, reflecting the regioselective nature of the reaction. Furthermore, under the optimized reaction conditions, cyclization was only feasible from substrates possessing a conjugated ene-yne side chain. Prenyl-substituted sulfide (2-(3-methylbut-2-enylthio) indole, 3aa) was found to be unreactive in these conditions. Overall, the new synthesis of functionalized tricyclic indole derivatives appears to be quite general in nature. Further chemical manipulations at positions 1, 5, and 6 can be conveniently performed using the strategically placed vital functional groups (nitrogen, methyl, bromine, and chlorine) at these sites in compound 3. These transformations could easily lead to analogues of biologically active carbazole alkaloids.⁷ In addition, the vinyl group of product 4 could also be exploited for the creation of new C−C bonds.¹³

The actual mechanism for the formation of product 4 might be complicated. A simplified mecha[nis](#page-6-0)m for the observed 6 endo-trig carbocyclization is outlined in Scheme 4. Typically, gold-catalyzed intramolecular hydroarylation of alkynes begins with nucleophilic attack at the alkynic/allenic carbon activated with cationic gold. 11 The apparent formation of an S-containing six-membered ring in 4a−k suggested that, prior to the hydroarylation ste[p,](#page-6-0) rearrangement of the conjugated ene-yne manifold present on sulfur takes place, as depicted in Scheme 4. In addition to 1D NOESY correlations, the creation of a spiro ring in products 4g−j provide further evidence in favor of the

Scheme 4. Proposed Mechanism for AuCl₃-Mediated Formation of Dihydrothiopyrano $[2,3-b]$ indoles

proposed rearrangement. We are currently investigating in detail the different aspects of the rearrangement of a conjugated ene-yne side chain under these conditions.

The structures of the indole-fused thiopyran products (4a− k) were confirmed using X-ray crystallographic data of compound $4k$, as shown in Figure 2.¹⁴

Figure 2. ORTEP representation of 4k (ellipsoids drawn at 40%).

In conclusion, we have developed an efficient methodology for the formation of a dihydrothiopyrano[2,3-b]indole skeleton starting from indoline-2-thione in two steps. First, a regioselective BF_3 etherate-catalyzed alkylation of thione using unsaturated alcohols results in 2-(2-alkenylthio)indole. Subsequently, a Au(III)-mediated reaction of 2-(2-alkenylthio) indole allowed access to an indole-fused dihydrothiopyran framework via sequential rearrangement of the ene-yne side chain and intramolecular hydroarylation at the C_3 position of the indole core.

EXPERIMENTAL SECTION

General. All reagents and solvents were used as supplied by commercial sources without further purification. Melting points were measured using a MEL-TEMP II apparatus and are uncorrected. Precoated fluorescent silica gel TLC plates were used to monitor the progress of the reactions. The $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra were obtained by a 300 MHz FT-NMR spectrometer. Chemical shifts of the ¹H NMR spectra are reported in parts per million (ppm) downfield from tetramethylsilane. IR spectra were recorded on an FT-IR spectrometer, and the values are expressed in cm[−]¹ . HR-ESIMS spectra were obtained using a micrOTOF mass analyzer and ESI positive ionization source.

General Procedure for the Syntheis of Compounds 3aa−ih. BF₃ etherate (2.01 mmol) was added dropwise to solution of appropriate indolin-2-thione (1.0 mmol) and alcohol (1.0 mmol) in dichloromethane (40 mL). The reaction mixture was stirred at room temperature (see Table 1 for reaction times). Excess dichloromethane was evaporated using rotavap after the reaction was complete. The residue was subjected to column chromatography (2% ethyl acetate in hexanes) to obtain com[p](#page-2-0)ounds 3aa−ih.

2-(3-Methylbut-2-enylthio)-1H-indole (3aa). Light yellow liquid (70%, 153 mg); ¹H NMR (300 MHz, CDCl₃) δ 8.09 (s, 1H), 7.57 (d, $J = 7.7$ Hz, 1H), 7.33 (d, J = 7.8 Hz, 1H), 7.20 (t, J = 7.5 Hz, 1H), 7.11 $(t, J = 7.3 \text{ Hz}, 1\text{H}), 6.65 \text{ (s, 1H)}, 5.36 \text{ (t, } J = 7.4 \text{ Hz}, 1\text{H}), 3.48 \text{ (d, } J =$ 7.9 Hz, 2H), 1.74 (s, 3H), 1.47 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 137.1, 136.8, 128.9, 128.6, 122.5, 120.2, 120.0, 119.9, 110.5, 109.0, 35.1, 25.7, 17.5; FT-IR ν_{max} (neat) 3396, 2965, 2912, 1439, 1337, 1226, 741 cm⁻¹; HRMS (m/z) calcd for C₁₃H₁₅NS 218.0998 [M + H]+ , found 218.0997.

2-(Cinnamylthio)-1H-indole (3ae). Colorless oil $(50\% , 134 \text{ mg})$; ¹H NMR (300 MHz, CDCl₃) δ 8.10 (s, 1H), 7.61 (d, J = 7.6 Hz, 1H), 7.37−7.11 (m, 8H), 6.74 (s, 1H), 6.35 (s, 2H), 3.64 (s, 2H); 13C NMR (75 MHz, CDCl3) δ 137.1, 136.7, 133.1, 128.7, 128.6, 128.2, 127.8, 126.4, 125.6, 122.7, 120.4, 120.1, 110.7, 109.7, 39.9; FT-IR ν_{max} (neat) 3393, 1487, 1435, 1337, 1228, 962, 736, 696 cm[−]¹ ; HRMS (m/z) calcd for $C_{17}H_{16}NS$ 266.0998 [M + H]⁺, found 266.0985.

(Z)-1-Methyl-2-(3-methylpent-2-en-4-ynylthio)-1H-indole (3eg). Pale yellow liquid (48%, 116 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.58 (d, J = 7.4 Hz, 1H), 7.23 (d, J = 7.9 Hz, 1H), 7.13 (d, J = 6.4 Hz, 2H), 6.70 (s, 1H), 5.81 (t, J = 7.6 Hz, 1H), 3.83 (s, 3H), 3.61 (d, J = 7.7 Hz, 2H), 3.00 (s, 1H), 1.84 (s, 3H); 13C NMR (75 MHz, CDCl3) δ 138.4, 133.1, 130.2, 127.4, 122.23, 120.7, 120.3, 119.6, 109.51, 109.4, 81.8, 81.6, 36.4, 29.9, 22.8; FT-IR ν_{max} (neat) 3276, 3049, 2921, 1706, 1610, 1457, 1323, 735 cm⁻¹; HRMS (m/z) calcd for C₁₅H₁₅NS 242.0998 [M + H]⁺, found 242.0986.

(Z)-5-Chloro-1-methyl-2-(3-methylpent-2-en-4-ynylthio)-1H-indole (3fg). Colorless liquid (54%, 150 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.53 (s, 1H), 7.18 (s, 2H), 6.61 (s, 1H), 5.79 (t, J = 7.5 Hz, 1H), 3.79 (s, 3H), 3.61 (d, J = 7.7 Hz, 2H), 2.97 (s, 1H), 1.83 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 136.7, 132.8, 132.0, 128.3, 125.3, 122.4, 121.0, 119.5, 110.4, 108.6, 81.8, 81.5, 36.1, 30.1, 22.8; FT-IR ν_{max} (neat) 3286, 2920, 1452, 1320, 788 cm[−]¹ ; HRMS (m/z) calcd for $C_{15}H_{14}C$ INS 276.0608 [M + H]⁺, found 276.0619.

(Z)-5-Bromo-1-methyl-2-(3-methylpent-2-en-4-ynylthio)-1H-indole (3gg). Liquid (53%, 171 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.69 (s, 1H), 7.30 (d, J = 8.5 Hz, 1H), 7.14 (d, J = 8.5 Hz, 1H), 6.61 $(s, 1H)$, 5.79 $(t, J = 7.5 \text{ Hz}, 1H)$, 3.78 $(s, 3H)$, 3.62 $(d, J = 7.7 \text{ Hz},$ 2H), 2.98 (s, 1H), 1.83 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 136.9, 132.8, 131.9, 128.9, 124.9, 122.6, 121.0, 112.9, 110.9, 108.5, 81.8, 81.5, 36.1, 30.0, 22.8; FT-IR ν_{max} (neat) 3283, 2919, 1714, 1606, 1455, 1318, 786 cm⁻¹; HRMS (m/z) calcd for C₁₅H₁₄BrNS 320.0103 [M + H]+ , found 320.0101.

(Z)-5-Fluoro-1-methyl-2-(3-methylpent-2-en-4-ynylthio)-1H-indole (3hg). Yellow liquid (56%, 146 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.22 (d, J = 9.1 Hz, 2H), 6.99 (t, J = 8.8 Hz, 1H), 6.64 (s, 1H), 5.80 (t, J = 7.6 Hz, 1H), 3.81 (s, 3H), 3.62 (d, J = 7.6 Hz, 2H), 2.99 (s, 1H), 1.84 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.9 (d, J_{C-F} = 234.5 Hz), 135.0, 132.9, 132.1, 127.5 (d, J_{C-F} = 10.2 Hz), 120.9, 110.6 (d, J_{C-F} = 26.4 Hz), 110.06 (d, J_{C-F} = 9.6 Hz), 108.9 (d, J_{C-F} = 4.8 Hz), 104.9 (d, JC−^F = 23.3 Hz), 81.8, 81.5, 36.2, 30.1, 22.8; FT-IR ν_{max} (neat) 3290, 2922, 1620, 1456, 1328, 844, 786 cm⁻¹; HRMS (*m*/ z) calcd for $C_{15}H_{14}F$ NS 260.0904 $[M + H]^+$, found 260.0892.

(Z)-6-Chloro-2-(3-methylpent-2-en-4-ynylthio)-1H-indole (3bg). Colorless liquid (53%, 140 mg); ¹H NMR (300 MHz, CDCl₃) δ 8.11 (s, 1H), 7.45 (d, $J = 8.3$ Hz, 1H), 7.31 (s, 1H), 7.07 (d, $J = 8.0$ Hz, 1H), 6.60 (s, 1H), 5.83 (t, J = 7.7 Hz, 1H), 3.69 (d, J = 7.7 Hz, 2H), 3.07 (s, 1H), 1.85 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 137.4, 133.6, 128.8, 128.4, 127.2, 121.0, 120.8, 120.7, 110.4, 109.1, 82.2, 81.6, 36.0, 22.84; FT-IR ν_{max} (neat) 3403, 3284, 2919, 1608, 1435, 1333, 807 cm⁻¹; HRMS (*m*/z) calcd for C₁₄H₁₂ClNS 262.0452 [M + H]⁺ , found 262.0448.

(Z)-5-Methyl-2-(3-methylpent-2-en-4-ynylthio)-1H-indole (3cg). Pale yellow liquid (50%, 122 mg); ¹H NMR (300 MHz, CDCl₃) δ 8.06 (s, 1H), 7.37 (s, 1H), 7.22 (d, $J = 8.1$ Hz, 1H), 7.04 (d, $J = 8.0$ Hz, 1H), 6.58 (s, 1H), 5.85 (t, $J = 7.5$ Hz, 1H), 3.69 (d, $J = 7.6$ Hz, 2H), 3.10 (s, 1H), 2.46 (s, 3H), 1.87 (s, 3H); 13C NMR (75 MHz, CDCl3) δ 135.6, 133.9, 129.3, 128.9, 127.8, 124.2, 120.4, 119.8, 110.2, 108.7, 82.2, 81.8, 36.2, 22.8, 21.5; FT-IR $ν_{max}$ (neat) 3392, 3281, 2918, 1442, 1325, 793 cm⁻¹; HRMS (*m*/z) calcd for C₁₅H₁₅NS 242.0998 [M $+ H$]⁺, found 242.0991.

(Z)-5-Bromo-2-(3-methylpent-2-en-4-ynylthio)-1H-indole (3dg). Pale yellow liquid (52%, 160 mg); ¹H NMR (300 MHz, CDCl₃) δ 8.19 (s, 1H), 7.68 (s, 1H), 7.22 (dd, J = 25.5, 8.0 Hz, 2H), 6.56 (s, 1H), 5.83 (t, $J = 7.3$ Hz, 1H), 3.70 (d, $J = 7.7$ Hz, 2H), 3.07 (s, 1H), 1.85 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 135.7, 133.5, 130.3, 129.7, 125.3, 122.6, 120.8, 113.2, 111.9, 108.1, 82.3, 81.6, 35.9, 22.8; FT-IR ν_{max} (neat) 3193, 2964, 1435, 1021, 793 cm⁻¹. HRMS (m/z) calcd for $C_{14}H_{13}BrNS$ 305.9952 [M + H]⁺, found 305.9947.

2-((3-Ethynylcyclohex-2-en-1-yl)thio)-1-methyl-1H-indole (3eh). Yellow liquid (58%, 145 mg); ¹H NMR (400 MHz, CDCl₃) δ 7.64−7.60 (m, 1H), 7.34 (dd, J = 8.2, 0.8 Hz, 1H), 7.28 (ddd, J = 8.2, 6.9, 1.2 Hz, 1H), 7.15 (ddd, $J = 8.0, 6.9, 1.2$ Hz, 1H), 6.80 (d, $J = 0.8$ Hz, 1H), 3.86 (s, 3H), 3.71−3.64 (m, 1H), 2.94 (s, 1H), 2.22−2.14 (m, 3H), 1.97−1.90 (m, 1H), 1.89−1.84 (m, 1H), 1.83−1.76 (m, 1H), 1.70−1.63 (m, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 138.4, 134.6, 129.9, 127.3, 123.0, 122.5, 120.5, 119.8, 110.4, 109.7, 84.49, 76.8, 46.4, 29.9, 29.0, 27.6, 19.2; HRMS (m/z) calcd for C₁₇H₁₈NS 268.1160 [M $+ H$]⁺, found 268.1154.

(Z)-2-(3-Methylpent-2-en-4-ynylthio)-1-phenyl-1H-indole (3ig). Pale yellow liquid (51%, 156 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.58−7.43 (m, 5H), 7.17 (s, 4H), 6.81 (s, 1H), 5.68 (t, J = 7.6 Hz, 1H), 3.48 (d, J = 7.7 Hz, 2H), 3.04 (s, 1H), 1.82 (s, 3H); ¹³C NMR $(75 \text{ MHz}, \text{CDCl}_3)$ δ 139.2, 137.4, 133.3, 131.6, 129.1, 128.5, 128.0, 127.7, 122.5, 120.4, 120.3, 120.0, 110.3, 109.1, 82.1, 81.7, 35.6, 22.8; FT-IR $\nu_{\rm max}$ (neat) 3278, 3046, 2920, 1496, 741, 695 cm⁻¹; HRMS (*m*/ z) calcd for $C_{20}H_{18}NS$ 304.1160 [M + H]⁺, found 304.1154.

2-(3-Ethynylcyclohex-2-enylthio)-1-phenyl-1H-indole (3fh). Pale yellow liquid (55%, 167 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.60– 7.41 (m, 6H), 7.17 (d, $J = 6.0$ Hz, 3H), 6.89 (s, 1H), 5.94 (d, $J = 3.8$ Hz, 1H), 3.46 (s, 1H), 2.88 (s, 1H), 1.79−1.43 (m, 6H); 13C NMR (75 MHz, CDCl3) δ 139.2, 137.3, 134.4, 130.7, 129.1, 128.6, 127.9, 127.5, 122.8, 120.5, 120.2, 111.0, 110.6, 84.4, 77.2, 45.6, 28.9, 27.4, 18.9; FT-IR ν_{max} (neat) 3291, 3011, 2942, 1472, 772 cm⁻¹; HRMS (m/z) calcd for C₂₂H₂₀NS 330.1316 [M + H]⁺, found 330.1311.

5-Bromo-2-((3-ethynylcyclohex-2-en-1-yl)thio)-1-methyl-1H-indole (3gh). Colorless oil (55%, 189 mg); ¹H NMR (400 MHz, CDCl₃) δ 7.73–7.71 (m, 1H), 7.34 (dd, J = 8.7, 1.9 Hz, 1H), 7.19 (d, J $= 8.7$ Hz, 1H), 6.70 (d, J = 0.8 Hz, 1H), 3.82 (s, 3H), 3.71–3.67 (m, 1H), 2.93 (s, 1H), 2.21−2.16 (m, 3H), 1.93−1.84 (m, 2H), 1.81−1.73 $(m, 1H)$, 1.72−1.63 $(m, 1H)$; ¹³C NMR (101 MHz, CDCl₃) δ 136.9, 134.3, 131.6, 128.8, 125.2, 123.3, 122.8, 113.1, 111.1, 109.3, 84.3, 76.9, 46.3, 30.1, 28.9, 27.6, 19.1; HRMS (m/z) calcd for C₁₇H₁₇BrNS 346.0256 [M + H]⁺, found 346.0260.

5-Chloro-2-(3-ethynylcyclohex-2-enylthio)-1-methyl-1H-indole (3ih). Colorless liquid (55%, 182 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.56 (s, 1H), 7.21 (s, 2H), 6.70 (s, 1H), 6.20 (s, 1H), 3.82 (s, 3H), 2.92 (s, 1H), 2.18 (s, 2H), 1.99−1.56 (m, 5H); 13C NMR (75 MHz, CDCl3) δ 136.7, 134.3, 131.7, 128.1, 125.6, 123.3, 122.7, 119.7, 110.7, 109.5, 84.3, 77.2, 46.3, 30.2, 29.0, 27.7, 19.2; FT-IR ν_{max} (neat) 3287, 2931, 1451, 1220, 790 cm⁻¹; HRMS (m/z) calcd for C₁₇H₁₇ClNS 302.0765 [M + H]⁺, found 302.0755.

General Procedure for the Syntheis of Compounds 4a−k. AuCl₃ (10 mol %, 0.016 mmol) was added to a solution of the appropriate indole sulfide (0.16 mmol) in toluene. The reaction mixture was stirred at 80 °C for 4−12 h. After the reaction was complete, excess toluene was eveporated using rotavap. The residue was subjected to column chromatography (2% ethyl acetate in hexanes) to obtain compounds 4a−k.

4,9-Dimethyl-4-vinyl-4,9-dihydrothiopyrano[2,3-b]indole (4a). Light yellow liquid (76%, 30.5 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.70 (d, J = 7.7 Hz, 1H), 7.33 (d, J = 7.8 Hz, 1H), 7.26 (t, J = 7.4 Hz, 1H), 7.18 (t, J = 7.1 Hz, 1H), 6.38 (dd, J = 16.8, 10.7 Hz, 1H), 6.19 (d, J = 9.9 Hz, 1H), 5.71 (d, J = 9.9 Hz, 1H), 5.28−5.18 (m, 2H), 3.71 (s, 3H), 1.82 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 144.8, 137.6, 129.8, 126.1, 124.9, 120.7, 119.5, 119.1, 112.3, 110.8, 109.3, 108.3, 41.2, 29.9, 27.7; FT-IR ν_{max} (neat) 2962, 2923, 1613, 1462, 1405, 1326, 992, 914, 735 cm⁻¹; HRMS (*m*/z) calcd for C₁₅H₁₅NS 242.0998 [M + H]⁺ , found 242.0987.

6-Chloro-4,9-dimethyl-4-vinyl-4,9-dihydrothiopyrano[2,3-b] indole (4b). Colorless liquid (70%, 31.5 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.50 (s, 1H), 7.11 (dd, J = 20.8, 8.6 Hz, 2H), 6.20 (dd, J = 17.0, 10.2 Hz, 1H), 6.09 (d, $J = 10.1$ Hz, 1H), 5.60 (d, $J = 9.8$ Hz, 1H), 5.12 (t, J = 12.4 Hz, 2H), 3.62 (s, 3H), 1.65 (s, 3H), ¹³C NMR (75 MHz, CDCl₃) δ 144.3, 136.0, 129.8, 127.1, 126.8, 124.8, 120.8, 118.8, 112.7, 110.5, 109.2, 108.9, 41.0, 30.1, 27.2; FT-IR $\nu_{\rm max}$ (neat) 2924, 1615, 1463, 1423, 1362, 1330, 995, 914, 710 cm[−]¹ ; HRMS (m/z) calcd for $C_{15}H_{14}C$ INS 276.0608 [M + H]⁺, found 276.0614.

6-Bromo-4,9-dimethyl-4-vinyl-4,9-dihydrothiopyrano[2,3-b] indole (4c). Pale yellow liquid $(67\%, 35 \text{ mg})$; ¹H NMR (300 MHz, CDCl₃) δ 7.65 (s, 1H), 7.21 (d, J = 8.7 Hz, 1H), 7.10 (d, J = 8.3 Hz, 1H), 6.20 (dd, J = 17.3, 10.5 Hz, 1H), 6.09 (d, J = 10.1 Hz, 1H), 5.60 $(d, J = 10.1 \text{ Hz}, 1H), 5.12 (t, J = 13.0 \text{ Hz}, 2H), 3.62 (s, 3H), 1.65 (s,$ 3H); ¹³C NMR (75 MHz, CDCl₃) δ 144.3, 136.3, 129.8, 127.7, 126.7, 123.4, 121.8, 112.8, 112.3, 110.5, 109.6, 108.9, 41.0, 30.1, 27.2; FT-IR νmax (neat) 2961, 2922, 1613, 1461, 1422, 1363, 1329, 992, 922, 713 cm⁻¹; HRMS (*m*/z) calcd for C₁₅H₁₄BrNS 320.0103 [M + H]⁺, found 320.0103.

6-Fluoro-4,9-dimethyl-4-vinyl-4,9-dihydrothiopyrano[2,3-b] indole (4d). Pale yellow liquid $(68%, 28.6 \text{ mg})$; ¹H NMR (300 MHz, CDCl₃) δ 7.21 (d, J = 10.1 Hz, 1H), 7.17–7.10 (m, 1H), 6.87 (t, J = 9.0 Hz, 1H), 6.20 (dd, $J = 17.0$, 10.5 Hz, 1H), 6.10 (d, $J = 9.8$ Hz, 1H), 5.61 (d, J = 9.9 Hz, 1H), 5.17−5.06 (m, 2H), 3.63 (s, 3H), 1.65 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.3 (d, J_{C−F} = 234.2 Hz), 144.3, 134.2, 129.7, 126.9, 126.3 (d, J_{C-F} = 10.2 Hz), 112.5, 110.6, 109.2, 108.8 (d, J_{C-F} = 3.2 Hz), 108.6 (d, J_{C-F} = 13.0 Hz), 104.6 (d, J_{C-F} = 24.4 Hz); FT-IR ν_{max} (neat) 2967, 2919, 1619, 1573, 1473, 1425, 1334, 989, 918, 712 cm⁻¹; HRMS (m/z) calcd for C₁₅H₁₄FNS 260.0904 [M + H]⁺, found 260.0916.

7-Chloro-4-methyl-4-vinyl-4,9-dihydrothiopyrano[2,3-b]indole (4e). Pale yellow liquid (66%, 27.8 mg); ¹H NMR (300 MHz, CDCl_3) δ 7.85 (s, 1H), 7.43 (d, J = 8.5 Hz, 1H), 7.25 (s, 1H), 7.01 (d, J = 8.4 Hz, 1H), 6.19 (dd, J = 17.4, 10.4 Hz, 1H), 6.07 (d, J = 9.9 Hz, 1H), 5.59 (d, J = 9.9 Hz, 1H), 5.11 (t, J = 12.3 Hz, 2H), 1.66 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 144.2, 136.7, 129.5, 129.4, 127.2, 124.9, 123.0, 120.2, 112.7, 110.9, 110.9, 110.2, 40.7, 27.0; FT-IR ν_{max} (neat) 2965, 2925, 1737, 1697, 1608, 1446, 1406, 1367, 1320, 991, 918, 719 cm⁻¹; HRMS (*m*/z) calcd for C₁₄H₁₂ClNS 262.0452 [M + H]⁺, found 262.0441.

6-Bromo-4-methyl-4-vinyl-4,9-dihydrothiopyrano[2,3-b]indole (4f). Yellow liquid (62%, 30.4 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.95 (s, 1H), 7.64 (s, 1H), 7.16 (dd, J = 17.1, 8.8 Hz, 2H), 6.19 (dd, J $= 16.9, 10.7$ Hz, 1H), 6.06 (d, J = 9.8 Hz, 1H), 5.59 (d, J = 10.3 Hz, 1H), 5.13 (t, J = 13.8 Hz, 2H), 1.65 (s, 3H); ¹³C NMR (75 MHz, CDCl3) δ 144.0, 135.0, 129.5, 128.1, 124.0, 123.8, 121.9, 113.0, 112.9, 111.56 110.8, 110.5, 40.6, 27.0; FT-IR ν_{max} (neat) 3193, 2964, 1435, 1042, 793 cm⁻¹; HRMS (*m*/z) calcd for C₁₄H₁₃BrNS 305.9952 [M + H]+ , found 305.9947.

9′-Methyl-9′H-spiro[cyclohex[2]ene-1,4′-thiopyrano[2,3-b] *indole] (4g).* Pale yellow liquid (60%, 24.5 mg); ¹H NMR (300 MHz, CDCl₃) δ 7.65 (d, J = 7.2 Hz, 1H), 7.26 (d, J = 7.9 Hz, 1H), 7.16 (t, J $= 6.8$ Hz, 1H), 7.07 (d, J = 6.9 Hz, 1H), 6.11 (d, J = 9.7 Hz, 1H), 5.99−5.84 (m, 3H), 3.66 (s, 1H), 2.37−2.16 (m, 3H), 2.06−1.69 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 137.7, 134.1, 129.7, 126.3, 125.9, 125.3, 120.6, 119.3, 118.9, 110.8, 110.3, 108.2, 38.7, 36.8, 29.9, 24.7, 18.6; FT-IR ν_{max} (neat) 3005, 2926, 1606, 1463, 1424, 1325, 727 cm⁻¹ ; HRMS (m/z) calcd for $C_{17}H_{17}NS$ 268.1154 $[M + H]^+$, found 268.1161.

6′-Chloro-9′-methyl-9′H-spiro[cyclohex[2]ene-1,4′-thiopyrano- [2,3-b]indole] (4h). Colorless liquid $(67%, 32.4 \text{ mg})$; ¹H NMR (300 MHz, CDCl₃) δ 7.55 (s, 1H), 7.18–7.01 (m, 2H), 6.08 (d, J = 8.6 Hz, 1H), 5.97−5.78 (m, 3H), 3.61 (s, 3H), 2.32−2.07 (m, 3H), 2.02−1.67 $(m, 3H)$; ¹³C NMR (75 MHz, CDCl₃) δ 136.1, 133.6, 129.8, 127.2, 126.5, 124.7, 120.7, 118.6, 110.6, 110.1, 109.1, 38.5, 36.7, 30.1, 24.6, 18.5; FT-IR ν_{max} (neat) 3018, 2930, 2850, 1609, 1462, 1421, 1329, 730 cm⁻¹; HRMS (*m*/z) calcd for C₁₇H₁₆ClNS 302.0765 [M + H]⁺, found 302.0766.

6′-Bromo-9′-methyl-9′H-spiro[cyclohex[2]ene-1,4′-thiopyrano- [2,3-b]indole] (4i). Pale yellow liquid $(61\%, 34.2 \text{ mg})$; ¹H NMR (300 MHz, CDCl₃) δ 7.71 (s, 1H), 7.20 (d, J = 8.6 Hz, 1H), 7.09 (d, J = 8.6 Hz, 1H), 6.09 (d, $J = 9.9$ Hz, 1H), 5.87 (dd, $J = 26.0$, 10.4 Hz, 3H), 3.62 (s, 3H), 2.29−2.06 (m, 3H), 2.01−1.67 (m, 3H); 13C NMR (75 MHz, CDCl₃) δ 136.4, 133.59, 129.9, 127.89 127.1, 126.5, 123.3, 121.7, 112.4, 110.63, 110.0, 109.5, 38.5, 36.8, 30.1, 24.6, 18.5; FT-IR ν_{max} (neat) 3022, 2929, 1462, 1237, 786 cm⁻¹; HRMS (*m*/z) calcd for $C_{17}H_{17}BrNS$ 346.0265 [M + H]⁺, found 346.0260.

9′-Phenyl-9′H-spiro[cyclohex[2]ene-1,4′-thiopyrano[2,3-b]indole] (4j). Pale yellow liquid (65%, 34.5 mg); ¹H NMR (300 MHz, $CDCl_3$) δ 7.74–7.65 (m, 1H), 7.60–7.51 (m, 2H), 7.47 (d, J = 7.2 Hz, 3H), 7.25−7.16 (m, 1H), 7.09 (dd, J = 6.0, 3.2 Hz, 2H), 6.05 (d, J = 9.9 Hz, 1H), 5.97−5.87 (m, 3H), 2.44−2.18 (m, 3H), 2.10−1.76 (m, 3H); 13C NMR (75 MHz, CDCl₃) δ 138.1, 136.7, 134.0, 129.5, 129.1, 128.0, 127.3, 126.7, 126.1, 125.5, 121.2, 119.7, 119.4, 111.9, 111.5, 109.4, 38.7, 36.6, 24.7, 18.6; FT-IR ν_{max} (neat) 3044, 2926, 1497, 1447, 736, 695 cm⁻¹; HRMS (*m*/z) calcd for C₂₂H₂₀NS 330.1316 [M + H]⁺ , found 330.1311.

4-Methyl-9-phenyl-4-vinyl-4,9-dihydrothiopyrano[2,3-b]indole (4k). Pale yellow liquid (69%, 34.5 mg); ¹H NMR (300 MHz, $CDCl₃$) δ 7.64 (dd, J = 6.0, 3.2 Hz, 1H), 7.54 (d, J = 6.8 Hz, 2H), 7.50–7.44 $(m, 3H)$, 7.19 (dt, J = 7.3, 3.7 Hz, 1H), 7.13–7.06 $(m, 2H)$, 6.34 (dd, J $= 17.3, 10.4$ Hz, 1H), 6.07 (d, J = 10.0 Hz, 1H), 5.61 (d, J = 10.0 Hz, 1H), 5.19 (dd, J = 13.9, 6.3 Hz, 2H), 1.77 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 144.7, 138.1, 136.7, 129.5, 129.2, 128.1, 127.4, 126.5, 125.2, 121.3, 119.8, 119.6, 112.4, 111.5, 110.8, 109.5, 41.2, 27.1; FT-IR ν_{max} (neat) 3045, 2962, 2922, 1594, 1497, 1447, 737, 696 cm⁻¹; HRMS (m/z) calcd for $C_{20}H_{18}NS$ 304.1160 $[M + H]^+$, found 304.1154.

■ ASSOCIATED CONTENT

6 Supporting Information

Copies of ${}^{1}H$ NMR and ${}^{13}C$ NMR spectra for products $3aa-$ 3ih and 4a−k, X-ray crystallographic data, and crystal structure of compound 4k. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/ jo5025943.

■ [AUTH](http://pubs.acs.org/doi/abs/10.1021/jo5025943)OR I[NFORMATION](http://pubs.acs.org)

Corresponding Author

*E-mail: mukundj@nipissingu.ca. Fax: 705-474-1947.

Notes

The auth[ors declare no competin](mailto:mukundj@nipissingu.ca)g financial interest.

■ ACKNOWLEDGMENTS

The authors gratefully acknowledge financial support provided by the Natural Sciences and Engineering Research Council of Canada (NSERC), Nipissing University, and the Canada Foundation for Innovation (CFI) to conduct this research.

■ REFERENCES

(1) (a) Wang, X.; Zhou, L.; Lu, W. Curr. Org. Chem. 2014, 18, 289. (b) Kitamura, T. Eur. J. Org. Chem. 2009, 1111. (c) Bandini, M.; Emer, E.; Tommasi, S.; Umani-Ronchi, A. Eur. J. Org. Chem. 2006, 3527. (d) Rao, V. K.; Shelke, G. M.; Tiwari, R.; Parang, K.; Kumar, A. Org. Lett. 2013, 15, 2190.

(2) For recent reviews, see: (a) Stephen, A.; Hashmi, K. Top. Organomet. Chem. 2013, 44, 143. (b) Boorman, T. C.; Larrosa, I. Chem. Soc. Rev. 2011, 40, 1910. (c) Bandini, M. Chem. Soc. Rev. 2011, 40, 1358. (d) Gilmore, K.; Alabugin, I. V. Chem. Rev. 2011, 111, 6513.

- (3) Yamamoto, Y. Chem. Soc. Rev. 2014, 43, 1575.
- (4) Blanchard, D.; Cameron, T. S.; Jha, M. Mol. Diversity 2013, 17, 827.
- (5) Majumdar, K. C.; Ponra, S.; Nandi, R. K. Tetrahedron Lett. 2012, 53, 1732.
- (6) (a) Takada, S.; Makisumi, Y. Chem. Pharm. Bull. 1984, 32, 872.
- (b) Takada, S.; Ishizuka, N.; Sasatani, T.; Makisumi, Y.; Jyoyama, H.; Hatakeyama, H.; Asanuma, F.; Hirose, K. Chem. Pharm. Bull. 1984, 32,
- 872. (c) Makisumi, Y.; Sasatani, T. United States Patent No. 4910318, March 20, 1990.
- (7) Schmidt, A. W.; Reddy, K. R.; Knoelker, H. Chem. Rev. 2012, 112, 3193.
- (8) (a) Jha, M.; Blunt, B. Tetrahedron Lett. 2009, 50, 6044. (b) Jha, M.; Chou, T. Y.; Blunt, B. Tetrahedron 2011, 67, 982. (c) Jha, M.; Guy, S.; Chou, T. Tetrahedron Lett. 2011, 52, 4337. (d) Jha, M.; Shelke, G.
- M.; Kumar, A. Eur. J. Org. Chem. 2014, 3334. (e) Jha, M.; Edmunds, M.; Lund, K.; Ryan, A. Tetrahedron Lett. 2014, 55, 5691. (f) Jha, M.;
- Davis, C.; Fazzari, J.; Vitali, M. Tetrahedron Lett. 2014, 55, 7043.
- (9) (a) Jha, M.; Enaohwo, O.; Marcellus, A. Tetrahedron Lett. 2009, 50, 7184. (b) Jha, M.; Enaohwo, O.; Guy, S. Tetrahedron Lett. 2011, 52, 684.
- (10) Berlin, J. M.; Goldberg, S. D.; Grubbs, R. H. Angew. Chem., Int. Ed. 2006, 45, 7591.
- (11) (a) Modha, S. G.; Kumar, A.; Vachhani, D. D.; Sharma, S. K.; Parmar, V. S.; Van der Eycken, E. V. Chem. Commun. 2012, 48, 10916. (b) Kumar, A.; Vachhani, D. D.; Modha, S. G.; Sharma, S. K.; Parmar, V. S.; Van der Eycken, E. V. Eur. J. Org. Chem. 2013, 2288.
- (12) (a) Qiu, Y.; Kong, W.; Fu, C.; Ma, S. Org. Lett. 2012, 14, 6198. (b) Liu, C.; Widenhoefer, R. A. Org. Lett. 2007, 9, 1935.
- (13) Qin, L.; Ren, X.; Lu, Y.; Li, Y.; Zhou, J. Angew. Chem., Int. Ed. 2012, 51, 5915.
- (14) Crystal structure of compound 4k: CCDC 1057877.